



Why do we use our soil test?



Understanding Superior's soil test

Superior's soil test compares each soil element against the ideal quantity and ratio for your crop and farm. The table below right is a typical example of the information we gain from this test, which is used to develop your customised Superior Fertiliser and Soil Management Plan:

Why do we only use our soil test?

Soil test results vary between laboratories due to different methods used and non-standardisation of protocols between laboratories.

For example, the standardisation of a soil test method involves defining parameters such as the solution mixed with the soil, how much solution is used, how long you shake a sample, what control or reference soil is used to calibrate instruments and compare your soil with, and so on.

Our fertiliser recommendations use rules specific to our soil test and would have to be recalibrated if we were to use an alternative laboratory. Even after recalibration there is no guarantee the outcomes would be consistent.

Our tests are processed by an independent laboratory in Missouri, United States, ensuring a faster turnaround, better value for money.



Soil Test

Sample Depth	10cm
C.E.C. (M.E.)	13.78
Soil pH	6.20
Organic Matter %	6.90
Nitrogen (N) Kg/Ha	123.20

C.E.C. tells us how much fertiliser your soil will hold

Base Saturation Percentage

	Desired %	Available%	Excess/(Deficit)%	Outcome
Calcium (Ca++) (60–72%)	68.00%	71.44%	3.44%	Good
Magnesium (Mg++) (10–12%)	12.00%	6.74%	(5.26%)	Very Deficient
Potassium (K+) (2.5–7.5%)	5.00%	3.48%	(1.52%)	Minimum
Sodium (Na+) (0.5–2.5%)	1.50%	1.14%	(0.36%)	OK
Total Base Saturation (T.B.S.)	86.50%	82.80%	(3.70%)	

These elements will influence pH and drive the structure of your soil

This compares what you have available against what is most desirable and an outcome

Expressed as kgs/ha

	Desired(kgs/ha)	Available(kgs/ha)	Excess/(Deficit)	Outcome
Calcium (Ca)	4,198	4,410	212	Good
Magnesium (Mg)	444	250	(195)	Very Deficient
Potassium (K)	620	419	(183)	Minimum
Sodium (Na)	106	81	(26)	OK
Phosphates (P2O5)	500	443	(57)	Good
Sulphate - S (SO4)	112	7	(105)	Very Deficient

Trace Elements

	Desired(kgs/ha)	Available(kgs/ha)	Excess/(Deficit)	Outcome
Boron (B)	3.4	4.5	1.1	Excellent
Iron (Fe)	1084.2	1,084.2		Excellent
Manganese (Mn)	300.0	96.3	(203.7)	Very Low
Copper (Cu)	11.2	0.2	(11.0)	Extremely Deficient
Zinc (Zn)	20.2	7.6	(12.5)	Very Deficient
Cobalt (Co)	3.4	1.3	(2.1)	Deficient
Molybdenum (Mo)				

Note: values in brackets denote deficiencies

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Phosphate Levels: Olsen versus Bray?

Superior uses both the Olsen and Bray phosphate (P) tests as suitability depends on whether the soil is alkaline or acidic.

The Bray-P2 test is a more reliable measure of plant available P on neutral (pH 7) or acid (pH < 7) soils, whereas the Olsen method is more reliable for in highly calcareous (lime containing) or alkaline soils (pH > 7.4).

It is therefore very important to use the correct test as using a Bray-P2 test on an alkaline soil tends to underestimate available P, whereas the same goes for using the Olsen on an acidic soil. Phosphorus (P) is an essential element for all plants, so it's important to know if the soil provides a sufficient concentration of P for optimum plant growth.

Since the majority of New Zealand soils are acidic, we commonly use the Bray-P2, but we still strike soils that have high pH's and therefore require an Olsen test. The Bray test has an advantage in that it measures any phosphate available from reactive phosphate rock (RPR) applications, which is important if you've been applying RPR. This is a major shortcoming of the Olsen test as it will not pick up reactive phosphate rock (RPR) residues.

Finally, it is important to remember that we use different scales for measuring phosphate levels when using Olsen and Bray tests. For example, a reading of 500kg/ha of available phosphate on an acid soil is excellent, whereas 125kg/ha is excellent on an alkaline soil.



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